

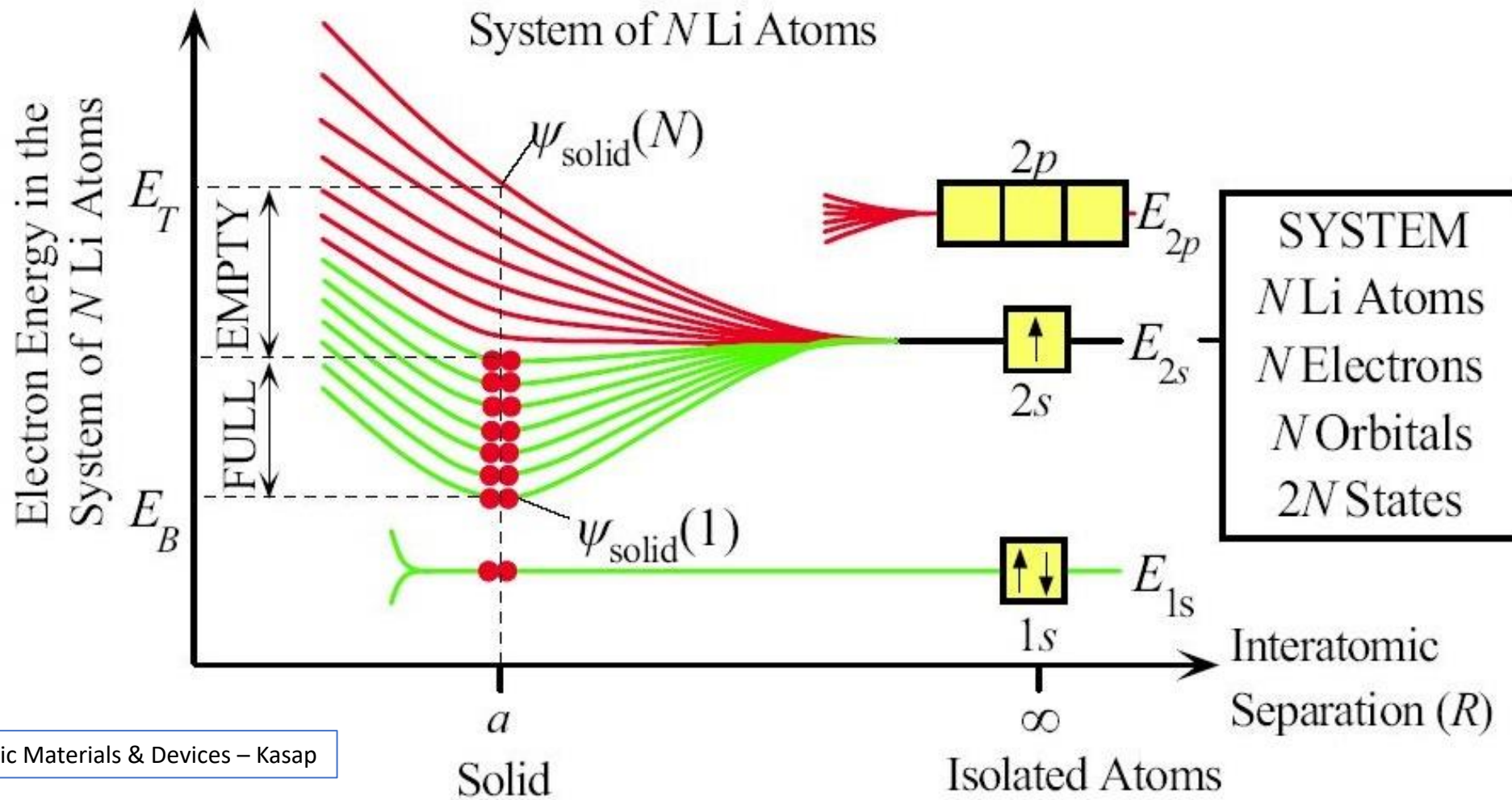
# Density of states

$$g(E) = \frac{1}{2\pi^2} \left( \frac{2m^*}{\hbar^2} \right)^{3/2} E^{1/2}$$

Dr Mohammad Abdur Rashid



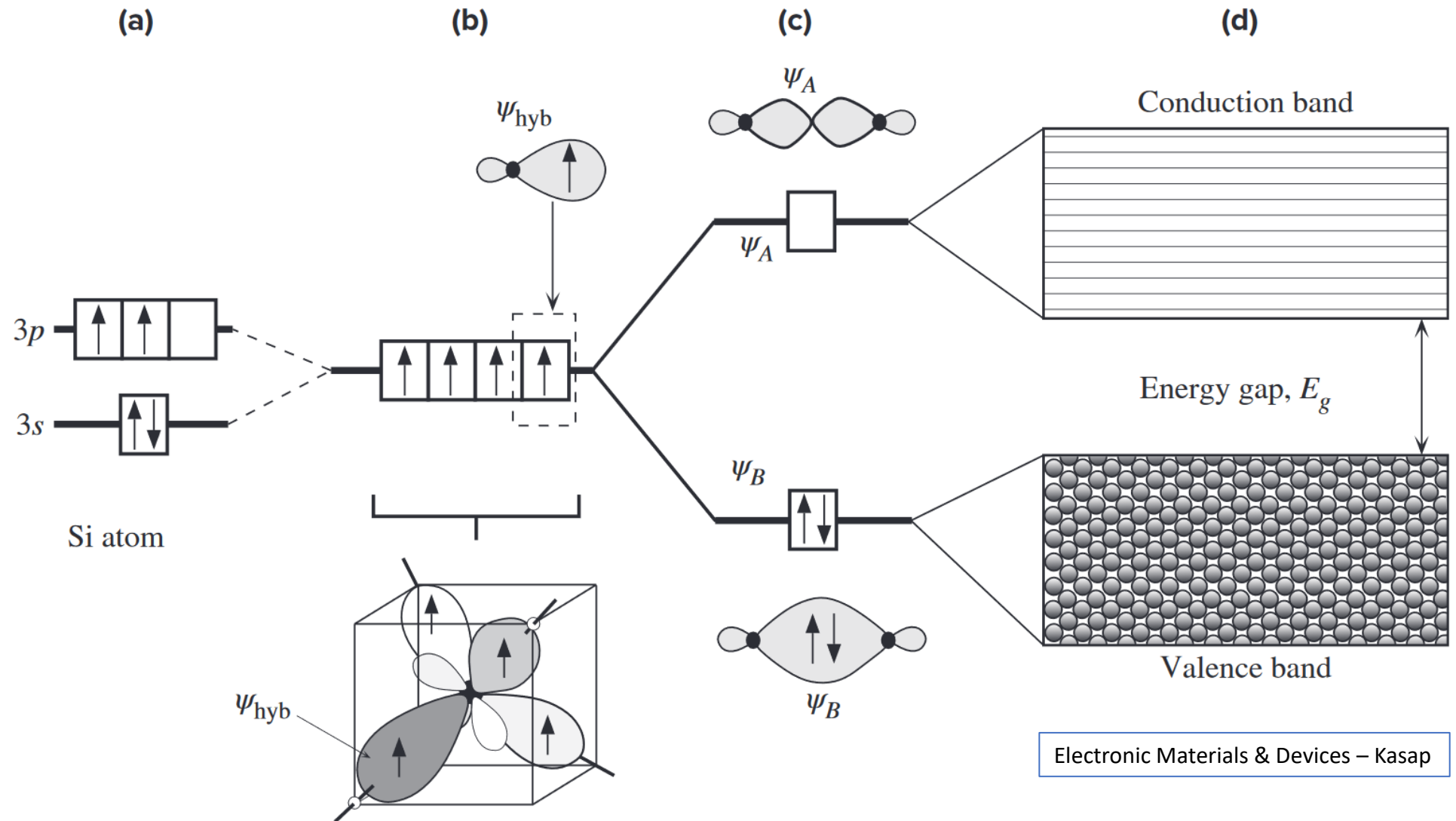
# Formation of solid - Lithium



Electronic Materials & Devices – Kasap

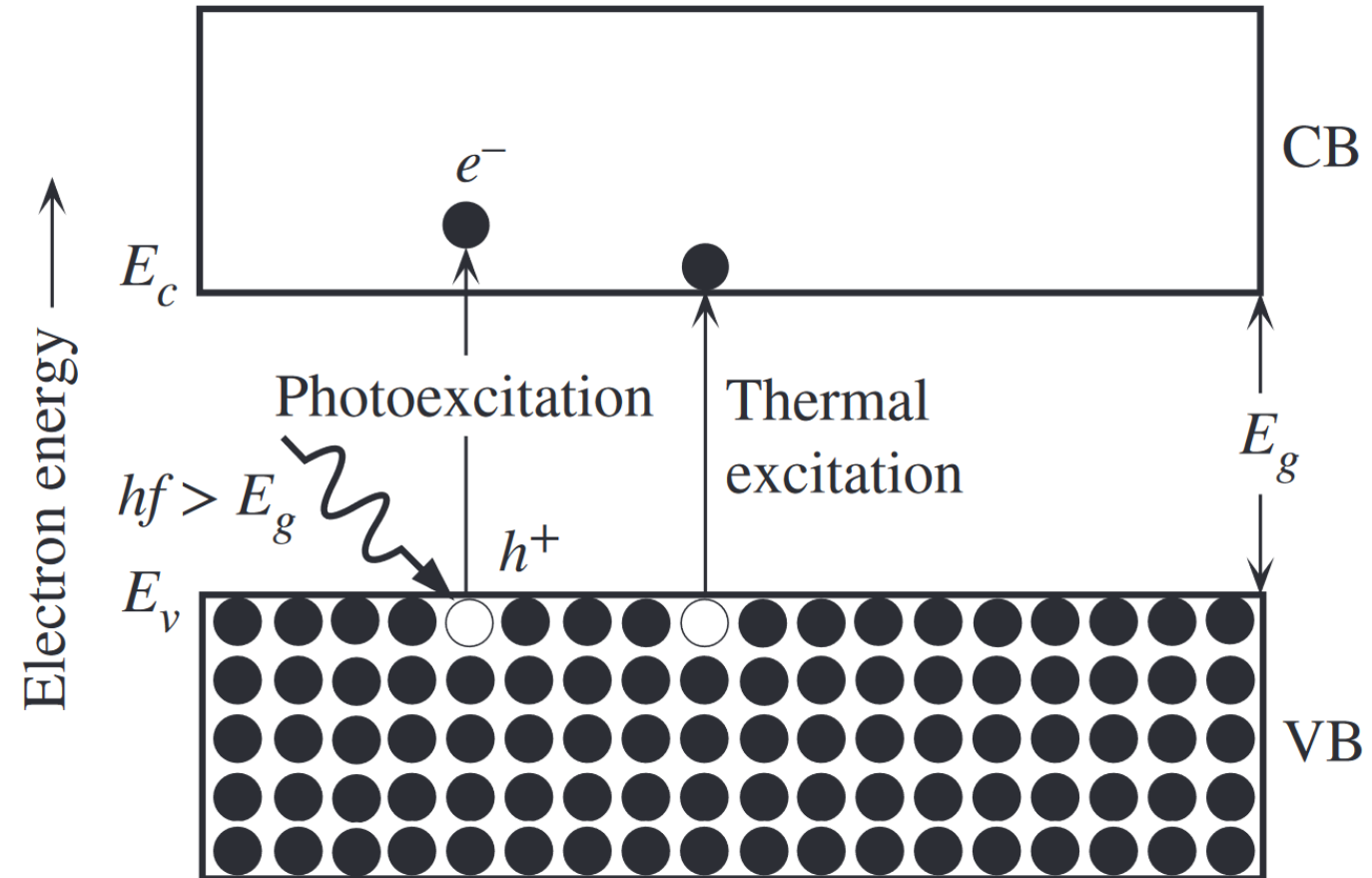


# Formation of energy band in silicon crystal



# Excitation of electrons from VB to CB

At room temperature

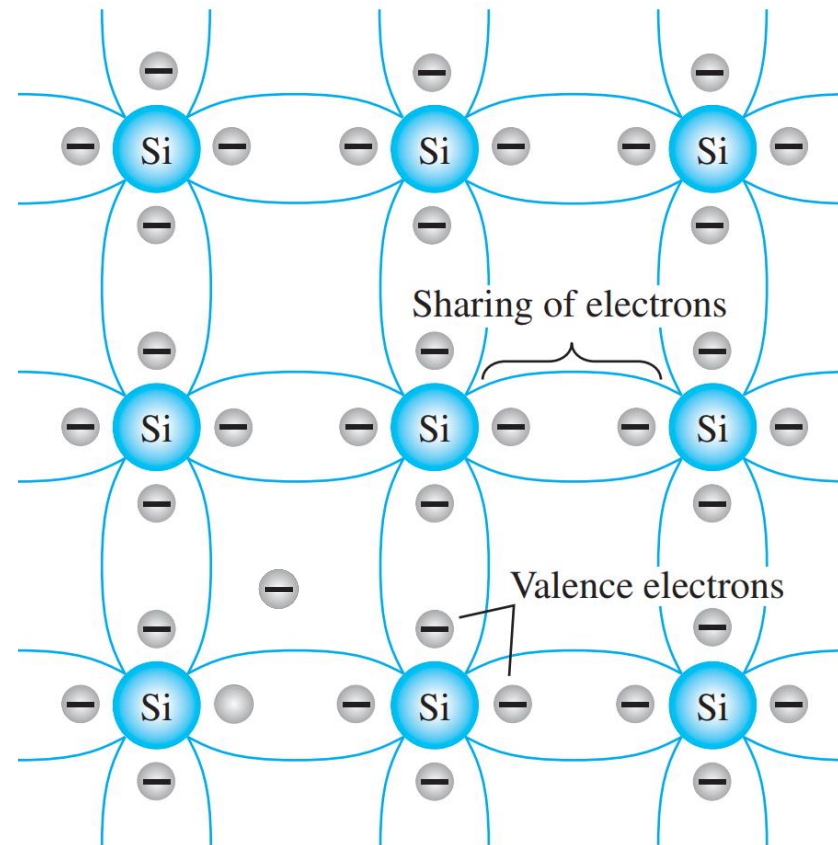


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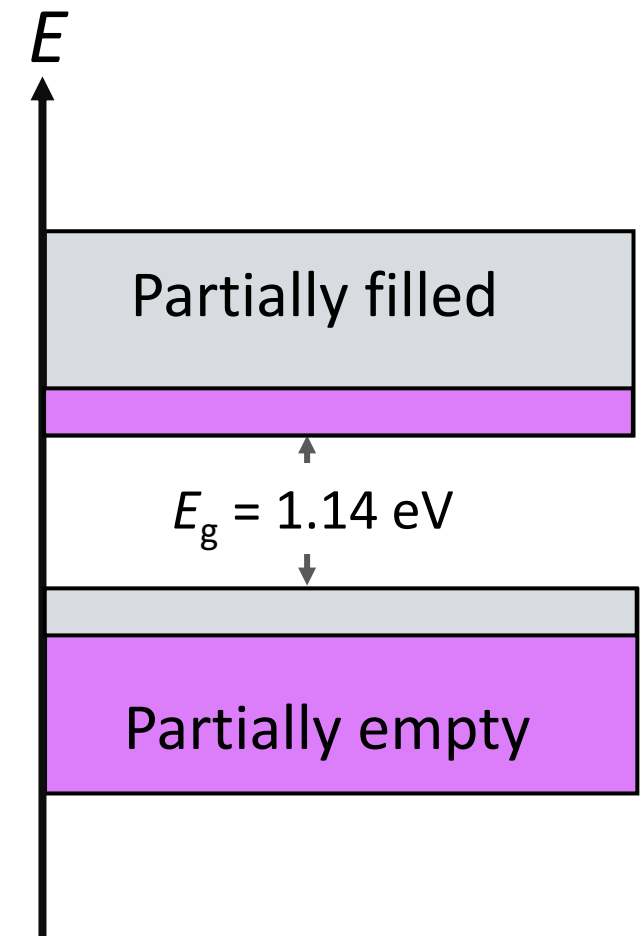


# Electron and Hole in intrinsic silicon

At room temperature there are approximately  $1.5 \times 10^{10}$  free carriers in  $1 \text{ cm}^3$  of *intrinsic* silicon.



Electronic Devices and Circuit Theory – Boylestad, Nashelsky



# Density of states

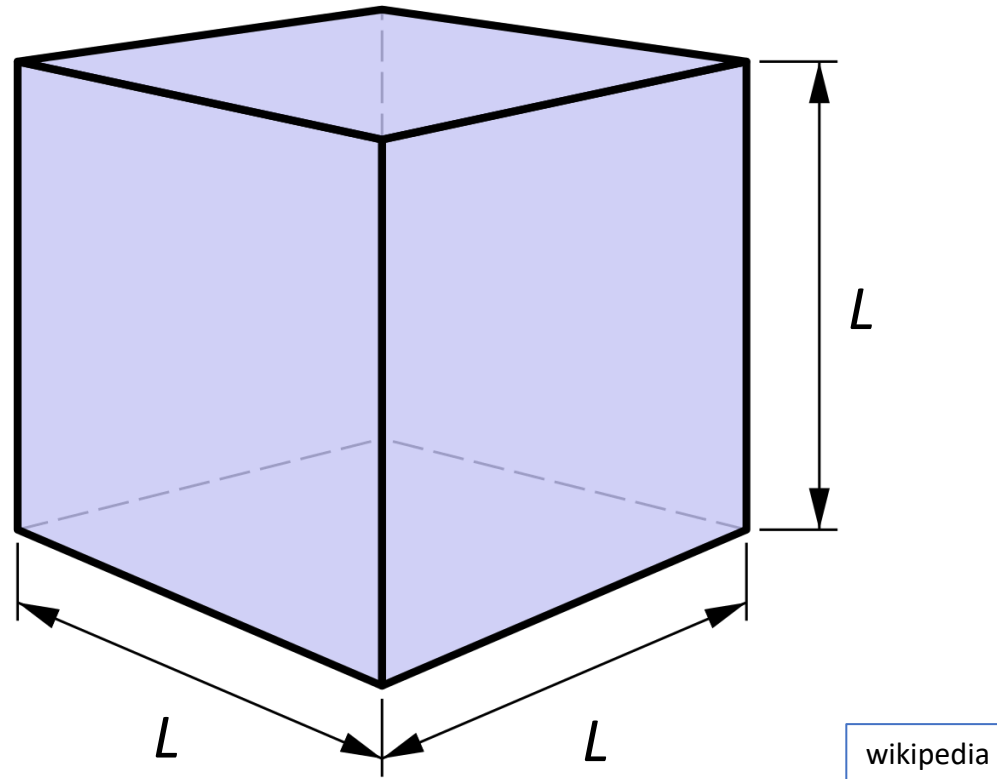
- Energy bands: A number of closely-spaced energy levels/states
- Density of states,  $g(E)$ :
  - $g(E)$  is the number of states per unit volume per unit energy range
  - $g(E) dE$  is the number of states per unit volume in the energy range  $(E, E + dE)$

$$g(E) \equiv \frac{\text{number of states in } \Delta E}{\Delta E \times \text{volume}}$$



# Density of states

Consider a cube of semiconductor crystal with length  $L$  on each side



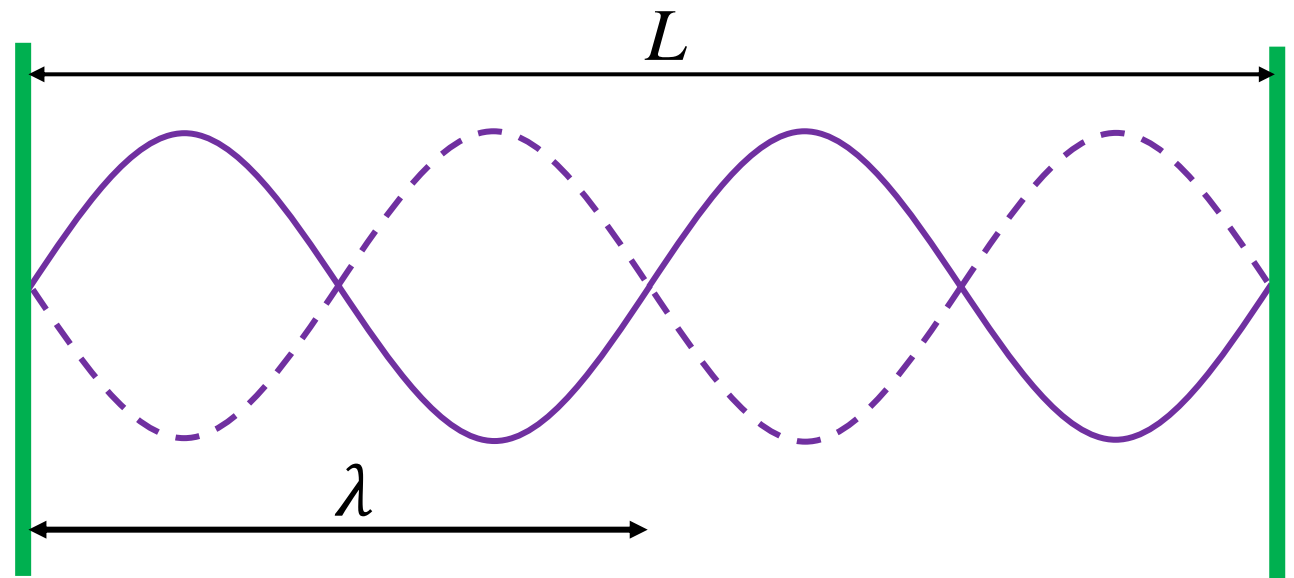
# Density of states

Consider a cube of semiconductor crystal with length  $L$  on each side

The electron waves in the crystal are standing waves

$$\lambda_x = \frac{L}{n_x}$$

$n_x$  is equal to 1, 2, 3 . . . .



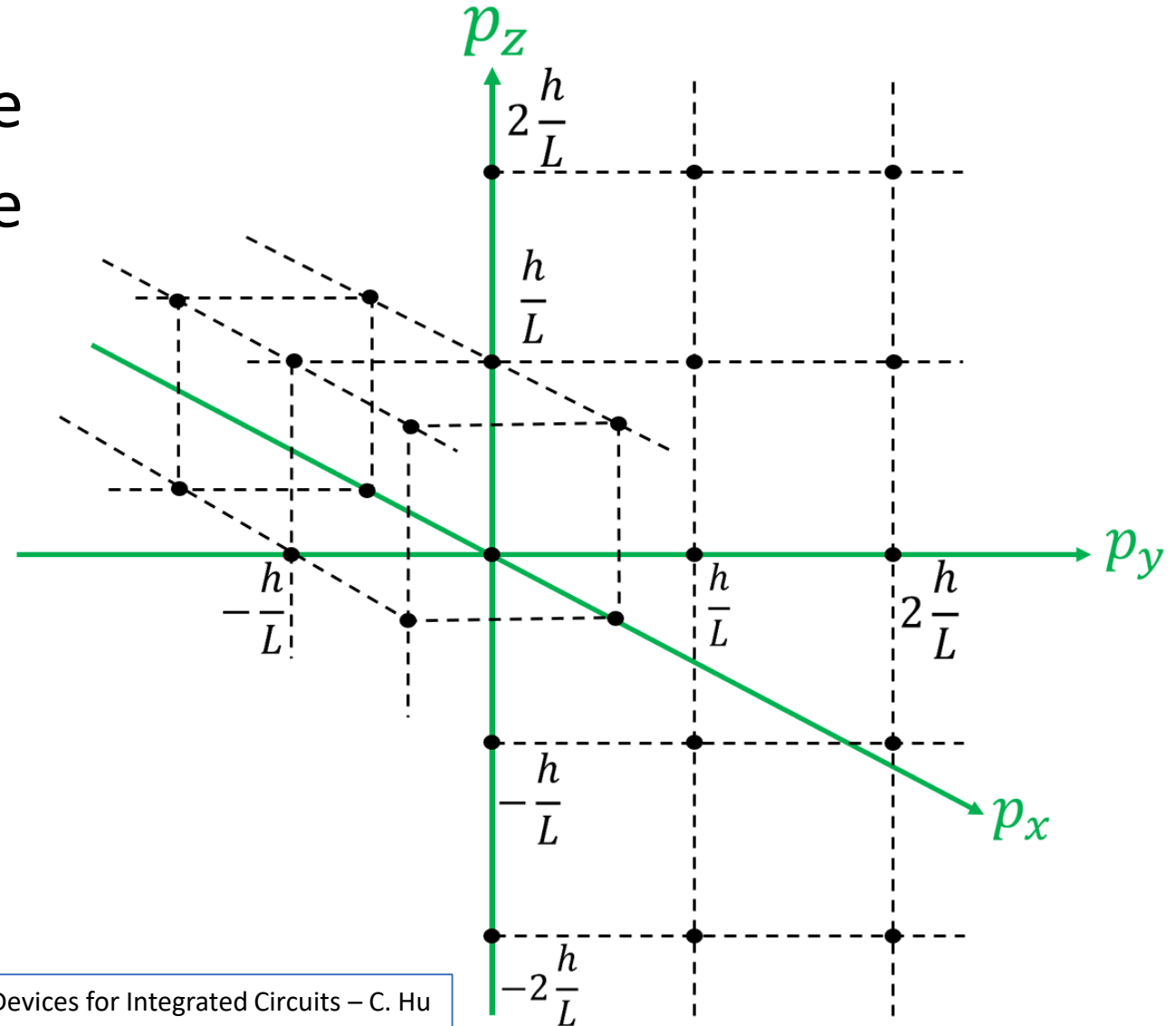


# Density of states

The wavelength is related to the electron momentum through the de Broglie relationship

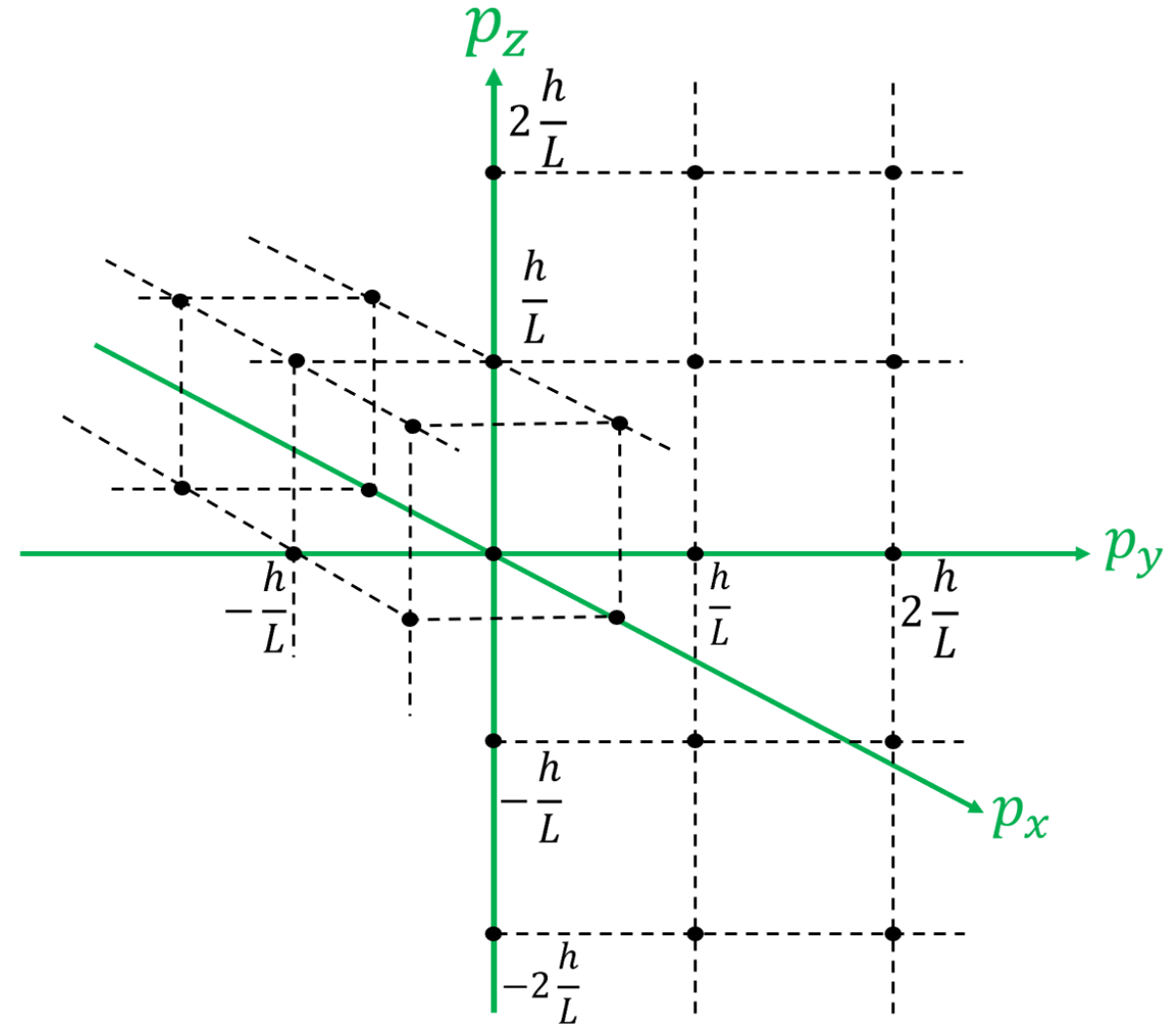
$$p_x = \pm \frac{h}{\lambda_x} = \pm \frac{n_x h}{L}$$

$n_x$  is equal to 1, 2, 3 . . . .



# Density of states

Allowed energy states occupy points separated from one another by  $h/L$  in  $p_x$ ,  $p_y$ , and  $p_z$ . There are two allowed states (the factor of 2 accounting for the two spin directions) for every cube of  $h^3/L^3$  volume in the momentum space. Each state therefore occupies a volume of  $h^3/2L^3$ .



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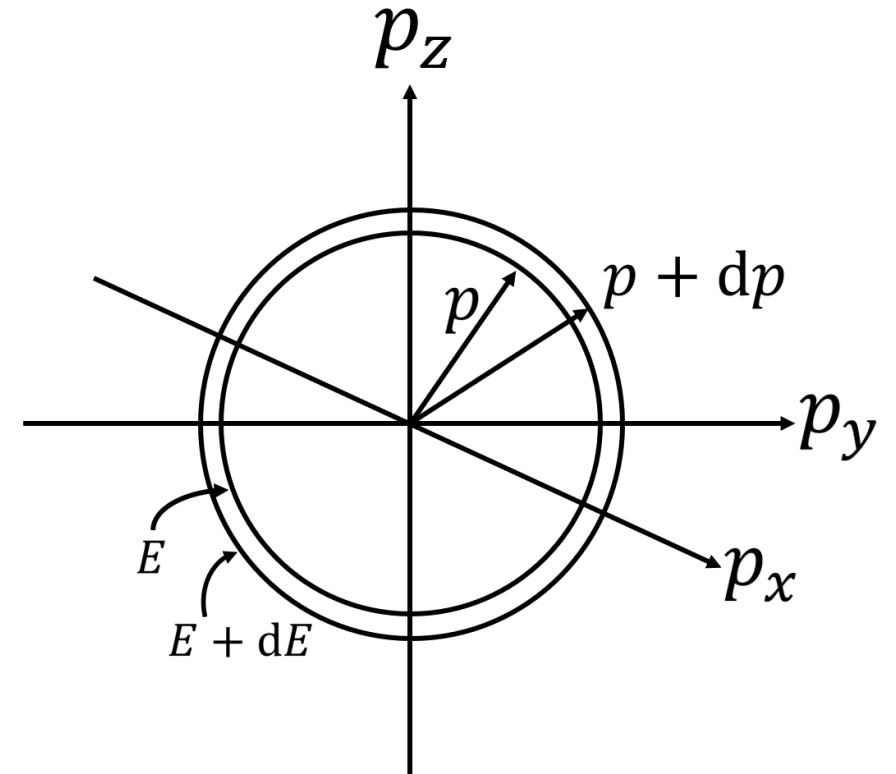


# Density of states

A sphere in the momentum space represents a constant total momentum,  $p$ , and therefore a constant kinetic energy,  $E$ .

$$E = \frac{p^2}{2m^*}$$

$m^*$  is the effective mass



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# Density of states

Kinetic energy:  $E = \frac{p^2}{2m^*}$

Derivative of  $E$  w.r.t.  $p$ :  $\frac{dE}{dp} = \frac{p}{m^*} = \frac{\sqrt{2m^*E}}{m^*} = \sqrt{\frac{2E}{m^*}}$

Rearranging:  $dp = \sqrt{\frac{m^*}{2E}} dE$



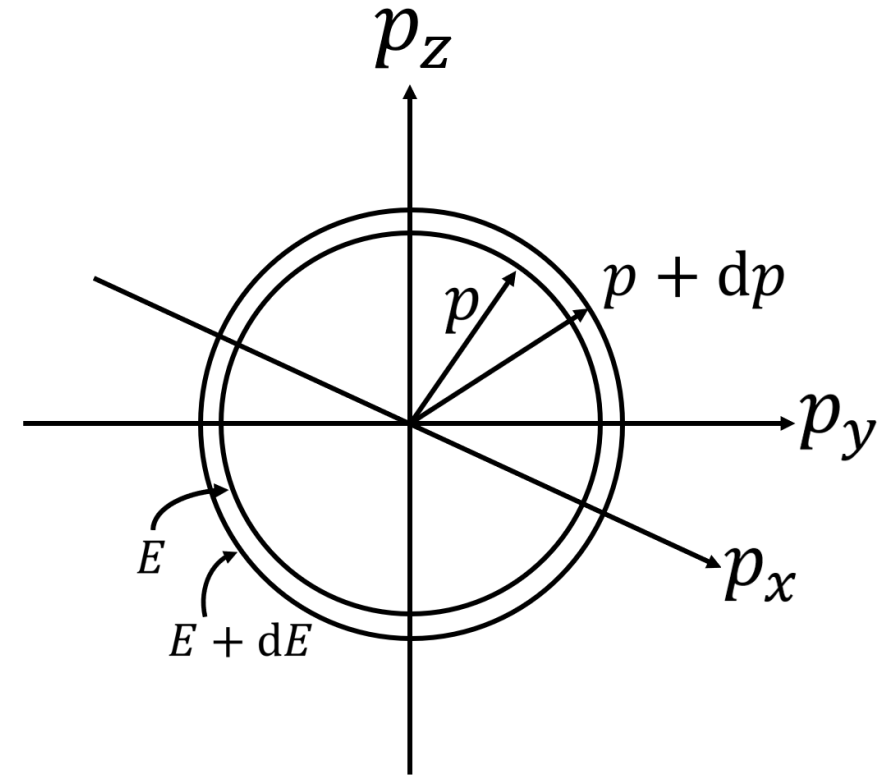
# Density of states

Two spheres that differ in energy  $dE$  have two radii that differ by  $dp$ :

$$dp = \sqrt{\frac{m^*}{2E}} dE$$

$$\text{Volume} = 4\pi p^2 dp = 4\pi(2m^* E) dp$$

$$= 8\pi m^* \sqrt{\frac{m^* E}{2}} dE$$



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# Density of states

The number of states contained in this shell between  $E$  and  $E + dE$  is

$$8\pi m^* \sqrt{\frac{m^* E}{2}} \times \frac{2L^3}{h^3} dE$$

The number of states per unit volume per unit energy is

$$g(E) = \frac{8\pi m^* \sqrt{2m^* E}}{h^3}$$

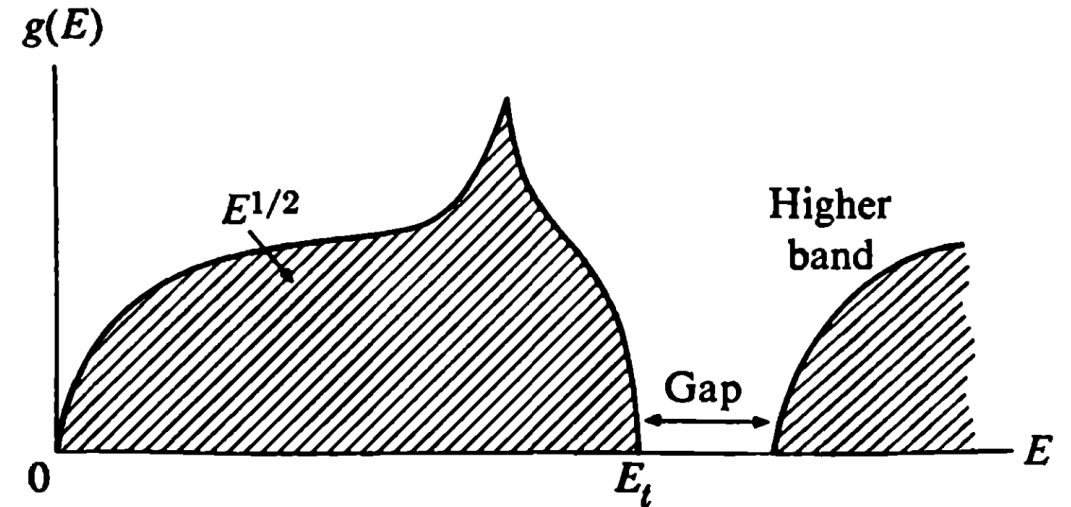


# Density of states

Using  $\hbar = h/2\pi$ , reduced Planck's constant, we have

$$g(E) = \frac{1}{2\pi^2} \left( \frac{2m^*}{\hbar^2} \right)^{3/2} E^{1/2}$$

$m^*$  is the effective mass



Elementary Solid State Physics – Ali Omar