

# Schrödinger Equation Operator in QM

Dr Mohammad Abdur Rashid



# de Broglie wave for a particle

The wave function or de Broglie wave for a free particle with momentum  $p$  and energy  $E$  is given by

$$\Psi(x, t) = e^{i(px - Et)/\hbar}$$

Since the wave function contains all the information about the system, it is often of interest to find appropriate operators to extract information from the wave function.



# Momentum operator $\hat{p}$

We take

$$\begin{aligned}\frac{\hbar}{i} \frac{\partial}{\partial x} \Psi(x, t) &= \frac{\hbar}{i} \frac{\partial}{\partial x} e^{i(px - Et)/\hbar} \\ &= \frac{\hbar}{i} \frac{ip}{\hbar} e^{i(px - Et)/\hbar} \\ &= p \Psi(x, t)\end{aligned}$$

where the  $p$  factor is just the momentum.



# Momentum operator $\hat{p}$

We thus identify the **momentum operator**  $\hat{p}$  as

$$\hat{p} \equiv \frac{\hbar}{i} \frac{\partial}{\partial x} = -i\hbar \frac{\partial}{\partial x}$$

and we have verified that acting on the wave function  $\Psi(x, t)$  for a particle of momentum  $p$  it gives  $p$  times the wave function:

$$\hat{p}\Psi(x, t) = -i\hbar \frac{\partial}{\partial x} \Psi(x, t) = p\Psi(x, t).$$



# Momentum operator $\hat{p}$

The momentum operator acts on wave functions, which are functions of space and time to give another function of space and time. Since  $\hat{p}$  acting on  $\Psi(x, t)$  gives a number ( $p$ , in fact) times  $\Psi(x, t)$  we say that  $\Psi(x, t)$  is an **eigenstate** of  $\hat{p}$ . We also say that  $\Psi(x, t)$  is a state of **definite momentum**.



# Energy operator $\hat{E}$

Let us now consider extracting the energy information from the free particle wave function. This time we must avail ourselves of the time derivative:

$$\begin{aligned}i\hbar \frac{\partial}{\partial t} \Psi(x, t) &= i\hbar \frac{\partial}{\partial t} e^{i(px - Et)/\hbar} \\ &= i\hbar \frac{-iE}{\hbar} e^{i(px - Et)/\hbar} \\ &= E\Psi(x, t).\end{aligned}$$



# Energy operator $\hat{E}$

Hence we define the total energy operator  $\hat{E}$  such that

$$\hat{E} = i\hbar \frac{\partial}{\partial t}.$$

$$\hat{E}\Psi(x, t) = i\hbar \frac{\partial}{\partial t} \Psi(x, t) = E\Psi(x, t)$$



# Eigenvalue Equation

$$\hat{p}\Psi(x, t) = p\Psi(x, t), \quad \hat{E}\Psi(x, t) = E\Psi(x, t)$$

(Operator)(function) = (constant factor) × (same function)

$$\text{(Operator)(eigenfunction) = (eigenvalue) × (eigenfunction)}$$

$$\text{Eigenvalue equation: } \hat{\Omega}\Psi = \omega\Psi$$





# Eigenvalue Equation

**Example:** Show that  $e^{ax}$  is an eigenfunction of the operator  $d/dx$ , and find the corresponding eigenvalue. Show that  $e^{ax^2}$  is not an eigenfunction of  $d/dx$ .

We need to operate on the function with the operator and check whether the result is a constant factor times the original function.



# Eigenvalue Equation

For  $\hat{\Omega} = \frac{d}{dx}$  and  $\Psi = e^{ax}$ :

$$\hat{\Omega}\Psi = \frac{d}{dx}e^{ax} = ae^{ax} = a\Psi$$

Therefore  $e^{ax}$  is indeed an eigenfunction of  $\frac{d}{dx}$ ,  
and its eigenvalue is  $a$ .



# Eigenvalue Equation

For  $\hat{\Omega} = \frac{d}{dx}$  and  $\Psi = e^{ax^2}$ :

$$\hat{\Omega}\Psi = \frac{d}{dx}e^{ax^2} = 2axe^{ax} = 2ax\Psi$$

which is not an eigenvalue equation of  $\hat{\Omega}$ . Even though the same function  $\Psi$  occurs on the right-hand side.



# Energy operator $\hat{E}$

$$\hat{p} \equiv \frac{\hbar}{i} \frac{\partial}{\partial x} = -i\hbar \frac{\partial}{\partial x}$$

$$\hat{E} = i\hbar \frac{\partial}{\partial t}$$

For a free particle the total energy is the kinetic energy and is given in terms of momentum such that

$$E = \frac{p^2}{2m}.$$



# Energy operator $\hat{E}$

We write

$$\begin{aligned} E\Psi &= \frac{p^2}{2m}\Psi = \frac{p}{2m}p\Psi = \frac{p}{2m}\hat{p}\Psi \\ &= \frac{p}{2m}(-i\hbar)\frac{\partial}{\partial x}\Psi \end{aligned}$$

Since  $p$  is a constant we can move the  $p$  factor on the last right-hand side close to the wave function and then replace it by the momentum operator.



# Energy operator $\hat{E}$

$$\begin{aligned} E\Psi &= \frac{1}{2m} (-i\hbar) \frac{\partial}{\partial x} p\Psi \\ &= \frac{1}{2m} (-i\hbar) \frac{\partial}{\partial x} (-i\hbar) \frac{\partial}{\partial x} \Psi \\ &= -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2}. \end{aligned}$$

$$\hat{E} \equiv \frac{\hat{p}^2}{2m} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}.$$



# Energy operator $\hat{E}$

For the free particle wave function  $\Psi(x, t) = e^{i(px - Et)/\hbar}$  we show that

$$\begin{aligned}\hat{E}\Psi(x, t) &= -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} e^{i(px - Et)/\hbar} \\ &= -\frac{\hbar^2}{2m} \left( \frac{ip}{\hbar} \right)^2 e^{i(px - Et)/\hbar} \\ &= \frac{p^2}{2m} \Psi(x, t) \\ &= E\Psi(x, t)\end{aligned}$$



# Free particle Schrödinger equation

$$i\hbar \frac{\partial}{\partial t} \Psi = E\Psi = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Psi$$

For a free particle the de Broglie wave function satisfies the differential equation:

$$i\hbar \frac{\partial}{\partial t} \Psi(x, t) = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \Psi(x, t)$$

The free-particle Schrödinger equation.

$$\Psi = e^{i(px - Et)/\hbar}$$

$$\hat{E} = i\hbar \frac{\partial}{\partial t}$$

$$\hat{E} \equiv \frac{\hat{p}^2}{2m} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$$





# Kinetic energy operator

Using the energy operator the free-particle Schrödinger equation can be written as:

$$i\hbar \frac{\partial}{\partial t} \Psi(x, t) = \hat{E} \Psi(x, t).$$

The kinetic energy operator:

$$\hat{T} \equiv -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$$



# Hamiltonian operator

If the particle is not free but rather is moving in some external potential  $V(x, t)$ , the total energy of the particle is the sum of kinetic and potential energies:

$$E = \frac{p^2}{2m} + V(x, t).$$

This suggests that the total energy operator should be:

$$\hat{E} = \frac{\hat{p}^2}{2m} + V(x, t) = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x, t).$$



# Hamiltonian operator

In case of a particle in a potential  $V(x, t)$  the Schrödinger equation is written as

$$i\hbar \frac{\partial}{\partial t} \Psi(x, t) = \left( -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x, t) \right) \Psi(x, t).$$

The energy operator  $\hat{E}$  is usually called the Hamiltonian operator  $\hat{H}$ , so one has

$$\hat{H} \equiv -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x, t).$$



# The Schrödinger equation

$$i\hbar \frac{\partial}{\partial t} \Psi(x, t) = \hat{H} \Psi(x, t)$$

For a particle quantum mechanical of mass  $m$  moving in a potential  $V(x, t)$  the Hamiltonian operator is

$$\hat{H} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x, t)$$



# Position operator

We introduce another very important operator: the **position operator**  $\hat{x}$  that acting on functions of  $x$  gives another function of  $x$  as follows:

$$\hat{x} f(x) \equiv x f(x).$$

$$\hat{x}^k f(x) \equiv x^k f(x)$$



# Schrödinger equation and wave function

$$i\hbar \frac{\partial}{\partial t} \Psi(x, t) = \hat{H} \Psi(x, t)$$

The Schrödinger equation has an explicit  $i$  on the left-hand side. This  $i$  shows that it is impossible to find a solution for real  $\Psi$ . If  $\Psi$  were real the right-hand side of the equation would be real but the left-hand side would be imaginary. Thus, the Schrödinger equation forces us to work with complex wave functions.



# The Schrödinger equation in 3D

$$i\hbar \frac{\partial}{\partial t} \Psi(\mathbf{r}, t) = \left( -\frac{\hbar^2}{2m} \nabla^2 + V(\mathbf{r}, t) \right) \Psi(\mathbf{r}, t)$$

The Laplacian operator:  $\nabla^2 \equiv \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$ .



# Operators in 3D

$$\hat{\mathbf{r}} \equiv (\hat{x}, \hat{y}, \hat{z})$$

$$\hat{\mathbf{p}} \equiv -i\hbar\nabla$$

$$\hat{T} \equiv -\frac{\hbar^2}{2m}\nabla^2$$

$$\hat{H} \equiv -\frac{\hbar^2}{2m}\nabla^2 + V(\mathbf{r}, t)$$

$$\nabla \equiv \left( \frac{\partial}{\partial x}, \frac{\partial}{\partial y}, \frac{\partial}{\partial z} \right)$$

$$\nabla^2 \equiv \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$$





# Thank You

**To receive notification of new video  
please subscribe to our channel.**

You may also let us know your comments.

